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Solution This problem gives us, for a hypothetical ion pair, values for r_0 (0.35 nm), E_0 6.13 eV, and n (10), and asks that we determine explicit expressions for attractive and repulsive energies of Equations 2.8 and 2.9. In essence, it is necessary to compute the values of A and B in these equations.

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for this metal at 900 C. Solution This problem is solved using two steps: (1) calculate the Callister Problem Solution - edugeneral.org Solution Heating to 700°C for 20 h the specimen in part (a) will transform the coarse pearlite and martensite to spheroidite. (c) Rapidly cool to 600°C (1110°F), hold for 4 s, rapidly cool to

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Solution This problem is solved using two steps: (1) calculate the total number of lattice sites in silver, N Ag, using Equation 4.2; and (2) multiply this number by fraction of lattice that are vacant, $2 \cdot 10^{-6}$. The parameter N Ag is related to the density, (ρ) , Avogadro's number (N A), and the atomic weight (A Ag =107.87 g/mol, from inside the

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Solution The average atomic weight of silicon (A) is computed by adding fraction-of-occurrence/atomic weight Si products for the three isotopes—i.e., using Equation 2.2. (Remember: fraction of occurrence is equal to the percent of occurrence divided by 100.) Thus. $A_{Si} = f_{28} A_{28} + f_{29} A_{29} + f_{30} A_{30}$

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Adapted from Fig. 9.3(a), Callister 7e. (Fig. 9.3(a) is adapted from Phase Diagrams of Binary Nickel Alloys, P. Nash (Ed.), ASM International, Materials Park, OH (1991). • 2 phases: L (liquid) ? (FCC solid solution) • 3 phase fields: L L + ? ? 0 20 40 60 80 100 wt% Ni 1000 1100 1200 1300 1400 1500 1600 T(°C) L (liquid) ? (FCC solid ...

Chapter 9: Phase Diagrams

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