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6. (c) 7. (d) 8. (b) 9. (a) 10. (c) NCERT Solutions for Class 11 Chemistry Chapter 6 HOTS Questions. Question 1. Why standard entropy of an elementary substance is not zero whereas the standard enthalpy of formation is taken as zero? Answer: A substance has a perfectly ordered arrangement only at absolute zero. Hence, entropy is zero only at ...

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Chapter 6 Equilibrium Chemistry 211 the reaction we monitor the mass of Ca^{2+} remaining in solution and the mass of CaCO_3 that precipitates, the result looks something like Figure 6.1. At the start of the reaction the mass of Ca^{2+} decreases and the mass of CaCO_3 increases. Eventually the reaction reaches a point after which there

Chapter 6

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