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Education. Laidler received his early education at Liverpool College. He received his BA (1934) and MA (1938) degrees from Trinity College, Oxford University. His MA was in the area of chemical kinetics under Cyril Norman Hinshelwood. He completed his PhD in 1940 from Princeton University, with a thesis entitled: The Kinetics of Reactions in Condensed and Heterogeneous Systems, under Henry Eyring.

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(i) Consider the rate law $r = k[H_2][I_2]$. If we substitute units into the equation, we obtain $(\text{mol dm}^{-3} \text{ s}^{-1}) = [k] (\text{mol dm}^{-3}) (\text{mol dm}^{-3})$ where the notation $[k]$ means 'the units of k '. We can rearrange this expression to find the units of the rate constant, k . $[k] = (\text{mol dm}^{-3} \text{ s}^{-1}) / (\text{mol dm}^{-3}) (\text{mol dm}^{-3}) = \text{mol}^{-1} \text{ dm}^3 \text{ s}^{-1}$

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