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 $\times 8.314 \times 10^5 \text{ (bar } \cdot \text{ m}^3 \text{ mol}^{-1} \text{ K}^{-1}) T$
 $P = \frac{V}{V_1} \times 10^3 \cdot (10^5 \cdot x) \cdot 10^6$
volume taken up by liquid Equating P
and P vap we have $x \cdot 18 \times 8.314 \times$
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An approximate solution can be found
if we combine Equations 1.4 and 1.5: 1
 $2 \text{ mV} \text{ e} \text{ kmolecular} \text{ } 2 \text{ } 3 \text{ kT} \text{ e} \text{ kmolecular}$
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