

## Hs Chemistry Pogil Activity Basic Stoichiometry Answers

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HS Chemistry POGIL Activity Page 2 Basic Stoichiometry Model 2  $2A + 3B \rightarrow 5C + 4D$  If 3 mol A react, how many mol C are produced? 3 mol A \* 5 mol C = 7.5 mol C 2 mol A 5. If 21.0 mol A are reacted, how many mol C are produced? Using dimensional analysis, show how you calculated your answer. 6.

HS Chemistry POGIL Activity Basic Stoichiometry

HS Chemistry POGIL Activity Topic: Measurement: Scientific Mathematics. Unit Dimensional Analysis Activity – Version 2. Why? In this activity we will see that it is possible to look at a situation from several points of view, or . to take measurements of that same situation using different units of measure. Every measurement has 2 . components: magnitude

Chemistry POGIL Activity «Activity

HS Chemistry POGIL Activity Topic: Stoichiometry. Basic Stoichiometry - KEY. Why? In this activity we will address the question: How do I convert between different chemical species in a given reaction? Model 1.  $2A + 3B \rightarrow 5C + 4D$ . 2 mol A produces 5 mol C 4 mol A produces 10 mol C. 3 mol B produces 4 mol D 6 mol B produces 8 mol D

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HS Chemistry POGIL Activity - Science Done Wright

HS Chemistry POGIL Activity Page 12 Basic Stoichiometry Model 3 Given the following equation:  $1 \text{ N}_2(\text{g}) + 3 \text{ H}_2(\text{g}) \rightarrow 2 \text{ NH}_3(\text{g})$  25. If 0.052 mol  $\text{N}_2$  are reacted, how many mol  $\text{NH}_3$  are formed? Using dimensional analysis, show how you calculated your answer.  
 $0.052 \text{ mol N}_2 \times 2 \text{ mol NH}_3 = 0.104 \text{ mol NH}_3$

Point out the important parts of their definitions and ...

HS Chemistry POGIL Activity Page 5 Basic Stoichiometry  $2 \text{ H}_2(\text{g}) + \text{ O}_2(\text{g}) \rightarrow 2 \text{ H}_2\text{O}(\text{g})$  14. Given the equation above, determine the number of moles of water produced when 5.2 g  $\text{O}_2$  are reacted. Make sure to show the dimensional analysis in your work. View full document.

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